

Percent Of Error Worksheet With Answers

Name: _____ Date: _____ Period: _____

Percent Error Practice 1.0 - Answer Key

Objective: Calculate Percent Error

Directions: Solve the following calculations using the percent error formula. Show your work:

1. In lab you recovered 24.3 grams of silver chloride but the reaction theoretically should have produced 27.4 grams. What is your percent error?

$$\% \text{ Error} = \frac{|\text{Accepted} - \text{Experimental}|}{\text{Accepted}} \times 100 = \frac{|27.4 \text{ g} - 24.3 \text{ g}|}{27.4 \text{ g}} \times 100 = \boxed{11.3\%}$$

2. You calculated from data that the speed of light is 3.4×10^8 m/s. You look up on Google that it is 3.0×10^8 m/s. What is your percent error?

$$\% \text{ Error} = \frac{|3.0 \times 10^8 \frac{\text{m}}{\text{s}} - 3.4 \times 10^8 \frac{\text{m}}{\text{s}}|}{3.0 \times 10^8 \frac{\text{m}}{\text{s}}} \times 100 = \boxed{13\%}$$

3. The density of iron is 7.87 g/mL. You mass a sample of iron and find its mass is 10.9 grams. You take the same sample and using the water displacement method find that the volume of water went from 22.30 mL to 23.60 mL. Based off your lab data, what would be the percent error for the density of the iron sample?

$$23.60 \text{ mL} - 22.30 \text{ mL} = 1.30 \text{ mL} \quad D = \frac{m}{V} = \frac{10.9 \text{ g}}{1.30 \text{ mL}} = 8.385 \text{ g/mL}$$
$$\% \text{ Error} = \frac{|\frac{10.9 \text{ g}}{1.30 \text{ mL}} - 7.87 \frac{\text{g}}{\text{mL}}|}{7.87 \frac{\text{g}}{\text{mL}}} \times 100 = \boxed{6.54\%}$$

4. What are the 2 possibilities for your fellow classmate's experimental value if their percent error was 12.5% and the actual value for the amount of heat released from a certain reaction is 24.0 J?

$$\% \text{ Error} = \frac{|24.0 \text{ J} - x|}{24.0 \text{ J}} \times 100 = 12.5\% \quad \frac{|24.0 \text{ J} - x|}{24.0 \text{ J}} = 0.125$$
$$|24.0 \text{ J} - x| = 3.00 \quad x = \boxed{27.0 \text{ J} \text{ and } 21.0 \text{ J}}$$

5. A perfect cube is 5.00 cm by 5.00 cm by 5.00 cm and has a mass of 196.5 g. What is the percent error of the density you calculate if the actual density of the cube is 1.48 g/cm³?

$$V = L \times W \times H = 5.00 \text{ cm} \times 5.00 \text{ cm} \times 5.00 \text{ cm} = 125 \text{ cm}^3$$
$$D = \frac{m}{V} = \frac{196.5 \text{ g}}{125 \text{ cm}^3} = 1.572 \text{ g/cm}^3$$
$$\% \text{ Error} = \frac{|\frac{1.48 \text{ g}}{1 \text{ cm}^3} - 1.572 \frac{\text{g}}{\text{cm}^3}|}{1.48 \frac{\text{g}}{\text{cm}^3}} \times 100 = \boxed{6.22\%}$$

Percent of Error Worksheet with Answers is a practical educational tool that helps students understand the concept of percent error, a crucial aspect of mathematics and applied sciences. Percent error is commonly used to measure the accuracy of experimental results compared to a known or accepted value. This article will provide a detailed overview of percent error, how to calculate it, its applications, and a sample worksheet with answers to enhance understanding.

Understanding Percent of Error

Percent error is defined as the difference between the experimental value and the accepted value, divided by the accepted value, and multiplied by 100 to express it as a percentage. It is a useful measure for evaluating the precision of measurements and is widely used in science, engineering, and various fields of research.

Formula for Percent Error

The formula for calculating percent error is:

$$\% \text{ Percent Error} = \left(\frac{\text{Experimental Value} - \text{Accepted Value}}{\text{Accepted Value}} \right) \times 100$$

Where:

- Experimental Value is the value obtained from measurements or experiments.
- Accepted Value is the true or theoretical value that is widely recognized.

Steps to Calculate Percent Error

1. Identify the Experimental Value: Determine the value obtained from your measurements or experiments.
2. Identify the Accepted Value: Find the known or accepted value for the measurement.
3. Subtract the Accepted Value from the Experimental Value: This gives you the difference.
4. Divide the Difference by the Accepted Value: This normalizes the error relative to the accepted value.
5. Multiply by 100: This converts the result into a percentage form.

Importance of Percent Error

Percent error is crucial in various fields for several reasons:

1. Accuracy Assessment: It helps in determining how close experimental results are to the accepted values, which is essential in scientific studies.
2. Quality Control: In manufacturing, percent error can indicate the quality and reliability of products.
3. Student Evaluation: Educators use percent error to teach students about precision and accuracy in measurements.

Applications of Percent Error

Percent error has numerous applications across different disciplines:

- Science Experiments: Used to evaluate the accuracy of experimental measurements.
- Engineering: Helps assess the performance of designs and materials against expected standards.
- Economics: Can be used to evaluate the accuracy of forecasts versus actual outcomes.
- Education: Assists teachers in providing feedback on students' experimental work.

Examples of Percent Error Calculation

To illustrate the concept further, let's explore a few examples of how to calculate percent error.

Example 1:

- Experimental Value: 98 grams
- Accepted Value: 100 grams

$$\text{Percent Error} = \left(\frac{98 - 100}{100} \right) \times 100 = \left(\frac{-2}{100} \right) \times 100 = -2\%$$

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Example 2:

- Experimental Value: 45.5 mL
- Accepted Value: 50 mL

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$$\text{Percent Error} = \left(\frac{45.5 - 50}{50} \right) \times 100 = \left(\frac{-4.5}{50} \right) \times 100 = -9\%$$

\]

Example 3:

- Experimental Value: 1.8 volts
- Accepted Value: 2 volts

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$$\text{Percent Error} = \left(\frac{1.8 - 2}{2} \right) \times 100 = \left(\frac{-0.2}{2} \right) \times 100 = -10\%$$

\]

Percent of Error Worksheet

Now that we have a solid understanding of percent error, let's create a worksheet that includes various problems for practice. Below is a sample worksheet with questions followed by answers.

Worksheet Questions

1. The density of a material is measured to be 8.5 g/cm³, while the accepted value is 8.96 g/cm³. Calculate the percent error.
2. A student measures the boiling point of water to be 98.6°C. What is the percent error if the accepted value is 100°C?
3. A length of a piece of wood is measured to be 3.2 meters, while the accepted value is 3.0 meters. Calculate the percent error.
4. The temperature in a lab is recorded as 22.5°C, but the accepted laboratory standard is 23.0°C. What is the percent error?
5. An experiment indicates that the mass of a sample is 45.7 grams, whereas the accepted value is 50 grams. Determine the percent error.

Answers to Worksheet Questions

1. Percent Error:

$$\left(\frac{8.5 - 8.96}{8.96} \right) \times 100 \approx -5.17\%$$

2. Percent Error:

$$\left(\frac{98.6 - 100}{100} \right) \times 100 = -1.4\%$$

3. Percent Error:

$$\left(\frac{3.2 - 3.0}{3.0} \right) \times 100 \approx 6.67\%$$

4. Percent Error:

$$\left(\frac{22.5 - 23.0}{23.0} \right) \times 100 \approx -2.17\%$$

5. Percent Error:

$$\left(\frac{45.7 - 50}{50} \right) \times 100 = -8.6\%$$

Conclusion

In summary, understanding how to calculate percent error is essential for students and professionals alike. The Percent of Error Worksheet with Answers provided in this article serves as a valuable resource for practicing and mastering this important concept. By regularly engaging with such worksheets, learners can enhance their mathematical skills and their ability to apply these concepts in real-world scenarios, fostering a deeper comprehension of measurement accuracy and reliability.

Frequently Asked Questions

What is a percent of error worksheet?

A percent of error worksheet is a tool used to calculate the difference between a measured or experimental value and a true or accepted value, expressed as a percentage of the true value.

How do you calculate percent of error?

Percent of error is calculated using the formula: Percent of Error = $\left| \frac{\text{Measured Value} - \text{Accepted Value}}{\text{Accepted Value}} \right| \times 100\%$.

What are some common applications of percent of error worksheets?

They are commonly used in science experiments, engineering, and quality control to assess accuracy and reliability of measurements.

What is the importance of understanding percent of error?

Understanding percent of error is crucial for evaluating the precision of measurements and ensuring the validity of experimental results.

Can a percent of error be negative?

No, percent of error is always expressed as a positive value, as it represents the magnitude of the difference between the measured and accepted values.

What is an example problem for a percent of error worksheet?

If the accepted value is 50 and the measured value is 45, the percent of error would be $|(45 - 50)| / 50 \times 100\% = 10\%$.

Are there any online tools for percent of error calculations?

Yes, there are several online calculators and tools available that can automatically compute percent of error when you input the measured and accepted values.

How can percent of error worksheets help students?

They help students practice calculations, understand measurement accuracy, and apply mathematical concepts in real-world scenarios.

What is a high percent of error indicative of?

A high percent of error indicates a significant discrepancy between the measured and accepted values, suggesting potential errors in measurement or methodology.

What is a low percent of error indicative of?

A low percent of error indicates that the measured value is close to the accepted value, suggesting higher accuracy in the measurement process.

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