

Half Life Problems Worksheet With Answers

Name: _____ Date: _____

Half - Life Practice

1. What is meant by half-life?
2. If you have 100 grams of a radioactive isotope with a half-life of 10 years:
 - a. How much of the isotope will you have left after 10 years?
 - b. How much of the isotope will you have left after 20 years?
 - c. How many half-lives will occur in 40 years?
3. The half-life of plutonium-239 is 24,300 years. If a nuclear bomb released 8 kg of this isotope, how many years would pass before the amount is reduced to 1 kg?
4. The half-life of radon-222 is 3.8 days. How much of a 100 gram sample is left after 15.2 days?
5. Carbon-14 has a half-life of 5,730 years. If a sample contained 70 mg originally, how much is left after 17,190 years?
6. The half-life of cobalt-60 is 5.26 years. If 50 grams are left after 15.78 years, how many grams were in the original sample?
7. The half-life of I-137 is 8.07 days. If 25 grams are left after 40.35 days, how many grams were in the original sample?
8. If 100 grams of Au-198 decays to 6.25 grams in 10.8 days, what is the half-life of Au-198?

Half life problems worksheet with answers is an essential tool for students and educators alike, particularly in the fields of chemistry and physics. Understanding half-life, the time required for a quantity to reduce to half its initial value, is crucial for grasping concepts related to radioactive decay, pharmacokinetics, and various applications in environmental science. This article will delve into half-life problems, provide a comprehensive worksheet, and offer answers to help learners solidify their understanding of the topic.

What is Half-Life?

Half-life refers to the time it takes for half of a given amount of a substance to decay or be eliminated. This concept is primarily associated with radioactive materials but can also apply to other processes, such as drug metabolism in the human body. The half-life of a substance can vary significantly

depending on its nature, and it is a crucial parameter in many scientific and practical applications.

Applications of Half-Life

Understanding half-life has several practical applications, including:

- **Radioactive Dating:** Determining the age of archaeological finds, such as fossils and ancient artifacts, using isotopes like carbon-14.
- **Medicine:** Calculating the correct dosage of medications based on their half-lives to ensure efficacy and safety.
- **Environmental Science:** Assessing the longevity of pollutants in ecosystems and their potential impact on health.
- **Nuclear Power:** Managing radioactive waste and understanding the decay of nuclear materials.

Common Half-Life Problems

Half-life problems can vary in complexity, but they often involve calculating the remaining quantity of a substance after a certain number of half-lives or determining the time it takes for a substance to decay to a specific amount. Here are some common types of half-life problems:

1. Remaining Quantity Calculation

In this type of problem, you calculate how much of a substance remains after a certain number of half-lives. The formula used is:

$$N = N_0 \left(\frac{1}{2}\right)^n$$

Where:

- N = remaining quantity
- N_0 = initial quantity
- n = number of half-lives

2. Time Calculation

These problems require determining the time it takes for a substance to decay to a specific amount. The formula is:

$$t = n \times t_{1/2}$$

Where:

- t = total time
- n = number of half-lives
- $t_{1/2}$ = half-life of the substance

Half Life Problems Worksheet

Below is a worksheet designed to help you practice half-life problems. Each problem is followed by its answer for self-assessment.

Worksheet Problems

1. A sample of a radioactive substance has an initial quantity of 80 grams. If its half-life is 3 years, how much will remain after 9 years?
2. A certain medication has a half-life of 4 hours. If a patient takes a dose of 200 mg, how much of the medication will be left in the patient's system after 12 hours?
3. A scientist has 50 mg of a radioactive isotope. If the half-life of the isotope is 5 days, how much will be left after 15 days?
4. If a substance's half-life is 10 years and you start with 160 grams, how long will it take for the substance to reduce to 20 grams?
5. A radioactive element has a half-life of 2 hours. If 1000 atoms of the element are present, how many atoms will remain after 6 hours?

Answers to the Worksheet Problems

1. Remaining Quantity Calculation:

- Initial quantity (N_0) = 80 grams
- Half-life ($t_{1/2}$) = 3 years
- Time elapsed = 9 years
- Number of half-lives (n) = $9 / 3 = 3$
- Remaining quantity (N) = $80 \left(\frac{1}{2}\right)^3 = 80 \times \frac{1}{8} = 10$ grams

2. Time Calculation:

- Initial quantity (N_0) = 200 mg
- Half-life ($t_{1/2}$) = 4 hours
- Time elapsed = 12 hours
- Number of half-lives (n) = $12 / 4 = 3$
- Remaining quantity (N) = $200 \left(\frac{1}{2}\right)^3 = 200 \times \frac{1}{8} = 25$ mg

3. Remaining Quantity Calculation:

- Initial quantity (N_0) = 50 mg
- Half-life ($t_{1/2}$) = 5 days
- Time elapsed = 15 days
- Number of half-lives (n) = $15 / 5 = 3$
- Remaining quantity (N) = $50 \left(\frac{1}{2}\right)^3 = 50 \times \frac{1}{8} = 6.25$ mg

4. Time Calculation:

- Initial quantity (N_0) = 160 grams
- Remaining quantity (N) = 20 grams
- Half-life ($t_{1/2}$) = 10 years
- Number of half-lives (n) = $\log(20/160) / \log(1/2) = 3$
- Total time (t) = 3×10 years = 30 years

5. Remaining Quantity Calculation:

- Initial quantity (N_0) = 1000 atoms
- Half-life ($t_{1/2}$) = 2 hours
- Time elapsed = 6 hours
- Number of half-lives (n) = $6 / 2 = 3$
- Remaining quantity (N) = $1000 \left(\frac{1}{2}\right)^3 = 1000 \times \frac{1}{8} = 125$ atoms

Conclusion

Understanding half-life is fundamental for students in many scientific disciplines. The use of a half-life problems worksheet with answers provides an excellent way to practice and reinforce this concept. By applying the formulas and solving various problems, students can enhance their grasp of half-life calculations, which are applicable in real-world scenarios such as medicine and environmental science. Whether you are a student preparing for exams or an educator looking for teaching resources, mastering half-life problems is a valuable skill that will serve you well in your academic journey.

Frequently Asked Questions

What is a half-life problems worksheet and what topics does it cover?

A half-life problems worksheet is a resource used in chemistry and physics that helps students practice calculations related to half-life, the time required for a quantity to reduce to half its initial value. It typically covers topics such as radioactive decay, exponential decay equations, and practical applications in science.

Where can I find half-life problems worksheets with answers?

Half-life problems worksheets with answers can be found on educational websites, online tutoring platforms, and resources like Teachers Pay Teachers. Many schools also provide printable worksheets

through their science departments.

How do I solve half-life problems effectively using a worksheet?

To solve half-life problems effectively, first identify the initial quantity and the half-life period provided in the problem. Use the formula $N = N_0 (1/2)^{(t/T)}$, where N is the remaining quantity, N_0 is the initial quantity, t is the total time, and T is the half-life. Practice with the worksheet by plugging in values and calculating the results step by step.

What are some common mistakes students make when solving half-life problems?

Common mistakes include misunderstanding the concept of half-life, miscalculating the number of half-lives elapsed, neglecting to convert time units, and confusing the initial quantity with the remaining quantity. Careful reading of the problem and consistent unit conversions can help avoid these errors.

Can half-life problems worksheets be used for exam preparation?

Yes, half-life problems worksheets are excellent for exam preparation as they provide practice with various types of problems that may be encountered on tests. Regular practice helps reinforce understanding and improve problem-solving skills related to half-life concepts.

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