

# Half Life Worksheet Extra Practice Answers

**Half Life Worksheet – Extra Practice**

*Answers Below*

- Fluorine-21 has a half life of approximately 5 seconds. What fraction of the original nuclei would remain after 1 minute?
- Iodine-131 has a half life of 8 days. What fraction of the original sample would remain at the end of 32 days?
- The half-life of chromium-51 is 28 days. If the sample contained 510 grams, how much chromium would remain after 56 days? How much would remain after 1 year? How much was present 168 days ago?
- If 20.0 g of a radioactive isotope are present at 1:00 PM and 5.0 g remain at 2:00 PM, what is the half life of the isotope?
- ~~The half life of Uranium-238 is 4.5 billion years and the age of earth is  $4.5 \times 10^9$  years. What fraction of Uranium-238 that was present when Earth was formed still remains?~~
- Chromium-48 decays. After 6 half-lives, what fraction of the original nuclei would remain?
- The half life of Iodine-125 is 60 days. What fraction of Iodine-125 nuclides would be left after 360 days?
- Titanium-51 decays with a half life of 6 minutes. What fraction of titanium would remain after one hour?
- ~~A medical institution requests 1 g of bismuth-214, which has a half life of 20 min. How many grams of bismuth-214 must be prepared if the shipping time is 2 h?~~
- ~~The half life of radium 226 is 1602 years. If you have 500 grams of radium today how many grams would have been present 9612 years ago?~~

*Handwritten solutions:*

1)  $\frac{1}{2^{12}} = \frac{1}{4096}$

2)  $\left(\frac{1}{2}\right)^4 = \frac{1}{16}$

3) 56 days:  $510 \cdot (0.5)^2 = 127.5g$

4)  $\frac{1}{2^6} = \frac{1}{64}$

5)  $\frac{1}{2^{12}} = \frac{1}{4096}$

6)  $\frac{1}{2^6} = \frac{1}{64}$

7)  $\frac{1}{2^6} = \frac{1}{64}$

8)  $\frac{1}{2^{10}} = \frac{1}{1024}$

*Handwritten table for problem 3:*

0	510	60
1	255	120
2	127.5	180

**Half life worksheet extra practice answers** are essential for students who are diving deep into the concepts of radioactive decay and half-life calculations. Understanding half-life is crucial not only in chemistry but also in fields such as biology, medicine, and environmental science. This article will provide a comprehensive overview of half-life, examples of calculations, and solutions to common problems found in practice worksheets. Additionally, we will discuss the significance of these concepts in real-world applications.

## Understanding Half-Life

Half-life is defined as the time required for half of the radioactive atoms in a sample to decay. This concept is fundamental in nuclear chemistry and

has broad applications in various scientific disciplines. When dealing with half-life calculations, it is important to grasp a few key concepts:

## Key Concepts of Half-Life

1. **Radioactive Decay:** This is the process by which an unstable atomic nucleus loses energy by emitting radiation. This decay occurs at a predictable rate characterized by the half-life.

2. **Exponential Decay:** The amount of substance remaining after each half-life decreases exponentially. This means that the decrease is not linear; rather, it diminishes more rapidly at first and slows down over time.

3. **Decay Constant ( $\lambda$ ):** This is a value that represents the probability of decay of a nucleus per unit time. It is related to half-life by the equation:

$$t_{1/2} = \frac{0.693}{\lambda}$$

4. **Remaining Quantity:** The amount of substance remaining after a certain number of half-lives can be calculated using the formula:

$$N = N_0 \left(\frac{1}{2}\right)^n$$

Where:

- $N$  = remaining quantity of the substance
- $N_0$  = initial quantity of the substance
- $n$  = number of half-lives that have passed

## Common Half-Life Problems

To solidify your understanding, let's explore some common problems associated with half-life worksheets. Providing extra practice answers can help students verify their understanding and improve their calculation skills.

### Sample Problems

1. **Problem 1:** A 100g sample of a radioactive substance has a half-life of 3 years. How much of the substance remains after 9 years?

Solution:

- Calculate the number of half-lives:  $(9 \text{ years}) \div 3 \text{ years/half-life} = 3$  half-lives.

- Remaining quantity:

$$N = 100g \left(\frac{1}{2}\right)^3 = 100g \times \frac{1}{8} = 12.5g$$

2. **Problem 2:** If a radioactive substance has a decay constant of 0.231 per year, what is its half-life?

Solution:

$$t_{1/2} = \frac{0.693}{0.231} \approx 3 \text{ years}$$

3. Problem 3: An isotope has a half-life of 5 days. If you start with 80mg, how much will remain after 15 days?

Solution:

- Number of half-lives:  $(15 \text{ days} \div 5 \text{ days/half-life}) = 3$  half-lives.

- Remaining quantity:

[

$$N = 80\text{mg} \left(\frac{1}{2}\right)^3 = 80\text{mg} \times \frac{1}{8} = 10\text{mg}$$

]

## Why Practice Half-Life Calculations?

Practicing half-life calculations through worksheets is vital for several reasons:

### Importance of Practice

- **Concept Reinforcement:** Regular practice helps reinforce the understanding of decay processes and the mathematical relationships involved.
- **Problem-Solving Skills:** Working through various problems enhances critical thinking and problem-solving skills essential in scientific fields.
- **Preparation for Exams:** Mastery of these concepts is often crucial for performance in exams and standardized tests where such calculations are included.
- **Real-World Applications:** Understanding half-life has practical implications in fields such as medicine (e.g., determining doses of radioactive tracers), archaeology (carbon dating), and environmental science (tracking pollutants).

## Tips for Success with Half-Life Worksheets

Here are some effective strategies for tackling half-life worksheets and improving your understanding of the topic:

### Study Strategies

1. **Review Basic Concepts:** Make sure you have a solid understanding of decay, half-life, and exponential functions before diving into complex problems.
2. **Use Visual Aids:** Graphs and diagrams can help visualize how substances decay over time. A graph showing the decay curve can be particularly helpful.
3. **Work with Peers:** Collaborating with classmates to solve half-life problems can enhance your understanding through discussion and shared insights.

4. **Practice Regularly:** Consistent practice with a variety of problems is key to mastering half-life calculations. Look for additional worksheets online or in textbooks.

5. **Utilize Online Resources:** Many educational platforms offer interactive worksheets and tutorials that can complement your learning.

## **Conclusion**

In summary, **half life worksheet extra practice answers** serve as a valuable resource for students aiming to master the concept of half-life in radioactive decay. By understanding the fundamental principles, solving practice problems, and employing effective study strategies, learners can enhance their comprehension of this essential scientific concept. Whether for academic purposes or real-world applications, a firm grasp of half-life will undoubtedly serve students well in their scientific endeavors.

## **Frequently Asked Questions**

### **What is a half-life worksheet and how is it used in practice?**

A half-life worksheet is an educational resource used to help students practice calculations related to half-life, the time required for a quantity to reduce to half its initial value, commonly used in chemistry and physics.

### **Where can I find extra practice answers for half-life worksheets?**

Extra practice answers for half-life worksheets can often be found in textbooks, educational websites, or online platforms like Khan Academy or educational forums where students and teachers share resources.

### **What types of problems are typically included in half-life worksheets?**

Half-life worksheets typically include problems that involve calculating the remaining quantity of a substance after a certain number of half-lives, determining the half-life from given data, and real-world applications like radioactive decay or pharmacokinetics.

### **How can I check my answers on half-life practice problems?**

You can check your answers by comparing them with answer keys provided with the worksheet, using online calculators, or validating your results through peer discussions or tutoring sessions.

### **Are there any online tools to assist with half-life**

## calculations?

Yes, there are several online calculators and apps that can assist with half-life calculations, offering features to input initial amounts, half-life duration, and the number of elapsed periods to calculate remaining quantities.

## Why is understanding half-life important in science?

Understanding half-life is crucial in science as it applies to various fields such as nuclear physics, radiometric dating, medicine (drug metabolism), and environmental science, helping to predict the behavior of substances over time.

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