

Ap Biology Unit 1 Cheat Sheet

<div>Cheatography</div> <div>AP Biology Unit 1: Biochemistry Cheat Sheet</div> <div>by hlewsey via cheatography.com/36676/cs/11528/</div>		
Properties of Water Polarity: - allows cohesion, adhesion, surface tension High Specific Heat: - resists temp change - high heat of vaporization - allows evaporative cooling (high energy particles vaporize) Universal Solvent: - dissolves hydrophilic/repels hydrophobic	Acids and Bases Acid increases H^+ in solution Base reduces H^+ in solution (accepts H^+ or donates OH^-) $pH = -\log[H^+]$ Buffer accepts/donates H^+ to stabilize pH Metabolism metabolic pathway a specific molecule that is altered in defined steps catalyzed by enzymes that result in a certain product catabolic pathway releases energy by breaking down complex molecules anabolic pathway consumes energy to build more complex molecules energy the capacity to cause change or rearrange matter	Free Energy Gibbs free energy- $\Delta G = \Delta H - T\Delta S$ energy in a temp/pressure constant system that can perform work exergonic has a net release of free energy ($-\Delta G$) endergonic absorbs free energy ($+\Delta G$) ATP ribose, adenine, 3 phosphate groups hydrolyzed to $ADP + P_i$ phosphorylated molecules = more reactive
Properties of Carbon tetravalence 4 bonds \rightarrow complex molecules hydrocarbons C and H, release energy structural isomer differs in covalent arrangement of atoms geometric isomer differs in spatial arrangements around double bonds enantiomer mirror image of 4 molecules attached to asymmetric carbon	Isomers - Structural  - Geometric  - Enantiomers 	Large Biomolecules Carbohydrates - monosaccharides held by glycosidic linkages - energy storage, structure, protection Lipids - fatty acids held by ester linkages - hydrophobic, saturated/unsaturated, hormones Nucleic Acids - nucleotides held by phosphodiester H-bonds - enable reproduction, controls protein synthesis Proteins - amino acids held by peptide bonds - follow varied instructions from DNA
Functional Groups -OH hydroxyl polar/hydrophilic -CH ₃ methyl nonpolar/hydrophobic - carboxyl polar/hydrophilic COOH -COH carbonyl polar/hydrophilic -NH ₂ amino polar/hydrophilic -SH sulfhydryl nonpolar/hydrophobic -PO ₄ phosphate polar/hydrophilic	Laws of Thermodynamics 1st Law of Thermodynamics energy can be transferred and transformed but not created or destroyed 2nd Law of Thermodynamics every energy transfer/transformation increases the entropy of the universe; a process must increase entropy to be spontaneous	Enzymes Enzymes these macromolecules act as catalysts that lower activation energy barriers by forming an induced fit w/substrate in the active site
<div>C</div> <div>By Hlewsey</div> <div>cheatography.com/hlewsey/</div>	Published 23rd April, 2017. Last updated 23rd April, 2017. Page 1 of 2.	Sponsored by CrosswordCheats.com Learn to solve cryptic crosswords! http://crosswordcheats.com

AP BIOLOGY UNIT 1 CHEAT SHEET IS AN ESSENTIAL RESOURCE FOR STUDENTS AIMING TO EXCEL IN THEIR ADVANCED PLACEMENT BIOLOGY COURSE. THIS UNIT SERVES AS THE FOUNDATION FOR UNDERSTANDING THE BIOLOGICAL PRINCIPLES THAT WILL BE EXPLORED THROUGHOUT THE COURSE. IT COVERS TOPICS SUCH AS THE CHEMISTRY OF LIFE, THE STRUCTURE AND FUNCTION OF MACROMOLECULES, AND THE BASIC PROPERTIES OF CELLS. IN THIS DETAILED ARTICLE, WE WILL BREAK DOWN THE KEY CONCEPTS, IMPORTANT TERMS, AND STRATEGIES FOR MASTERING UNIT 1 OF AP BIOLOGY.

1. THE CHEMISTRY OF LIFE

UNDERSTANDING THE CHEMISTRY OF LIFE IS CRUCIAL FOR ANY BIOLOGY STUDENT. THIS SECTION COVERS THE BASIC ELEMENTS AND COMPOUNDS THAT FORM THE BUILDING BLOCKS OF LIFE.

1.1. ELEMENTS AND COMPOUNDS

- KEY ELEMENTS: THE FOUR MAIN ELEMENTS THAT MAKE UP 96% OF LIVING MATTER ARE:

1. CARBON (C)
2. HYDROGEN (H)
3. OXYGEN (O)
4. NITROGEN (N)

- TRACE ELEMENTS: ELEMENTS SUCH AS IRON (Fe), IODINE (I), AND ZINC (Zn) ARE ALSO VITAL FOR VARIOUS BIOLOGICAL FUNCTIONS.

- COMPOUNDS: A COMPOUND IS A SUBSTANCE FORMED WHEN TWO OR MORE ELEMENTS CHEMICALLY BOND TOGETHER. COMMON BIOLOGICAL COMPOUNDS INCLUDE:

- WATER (H₂O)
- CARBOHYDRATES
- PROTEINS
- NUCLEIC ACIDS

1.2. WATER PROPERTIES

WATER IS ESSENTIAL FOR LIFE, AND ITS UNIQUE PROPERTIES ARE CRITICAL FOR BIOLOGICAL PROCESSES.

- COHESION AND ADHESION:
 - COHESION REFERS TO THE ATTRACTION BETWEEN MOLECULES OF THE SAME SUBSTANCE (E.G., WATER MOLECULES).
 - ADHESION REFERS TO THE ATTRACTION BETWEEN DIFFERENT SUBSTANCES (E.G., WATER AND PLANT CELL WALLS).
- HIGH SPECIFIC HEAT: WATER CAN ABSORB A LOT OF HEAT BEFORE ITS TEMPERATURE RISES, WHICH HELPS REGULATE TEMPERATURES IN ORGANISMS.
- SOLVENT OF LIFE: WATER IS KNOWN AS THE "UNIVERSAL SOLVENT" BECAUSE IT CAN DISSOLVE MANY SUBSTANCES, FACILITATING CHEMICAL REACTIONS IN CELLS.

2. BIOLOGICAL MACROMOLECULES

MACROMOLECULES ARE LARGE MOLECULES ESSENTIAL FOR LIFE, AND THEY INCLUDE CARBOHYDRATES, LIPIDS, PROTEINS, AND NUCLEIC ACIDS.

2.1. CARBOHYDRATES

- STRUCTURE: CARBOHYDRATES CONSIST OF CARBON, HYDROGEN, AND OXYGEN, TYPICALLY IN A RATIO OF 1:2:1.
- TYPES:
 1. MONOSACCHARIDES (E.G., GLUCOSE, FRUCTOSE)
 2. DISACCHARIDES (E.G., SUCROSE, LACTOSE)
 3. POLYSACCHARIDES (E.G., STARCH, GLYCOGEN, CELLULOSE)
- FUNCTIONS:
 - ENERGY STORAGE (E.G., STARCH IN PLANTS, GLYCOGEN IN ANIMALS)
 - STRUCTURAL SUPPORT (E.G., CELLULOSE IN PLANT CELL WALLS)

2.2. LIPIDS

- STRUCTURE: LIPIDS ARE HYDROPHOBIC MOLECULES THAT INCLUDE FATS, OILS, AND STEROIDS.
- TYPES:
 1. TRIGLYCERIDES (FATS AND OILS)
 2. PHOSPHOLIPIDS (CELL MEMBRANES)
 3. STEROIDS (CHOLESTEROL, HORMONES)
- FUNCTIONS:
 - LONG-TERM ENERGY STORAGE
 - INSULATION AND PROTECTION
 - CELL MEMBRANE STRUCTURE

2.3. PROTEINS

- STRUCTURE: PROTEINS ARE MADE OF AMINO ACIDS LINKED BY PEPTIDE BONDS.
- LEVELS OF STRUCTURE:
 1. PRIMARY STRUCTURE (SEQUENCE OF AMINO ACIDS)
 2. SECONDARY STRUCTURE (ALPHA HELICES AND BETA SHEETS)
 3. TERTIARY STRUCTURE (3D FOLDING)
 4. QUATERNARY STRUCTURE (MULTIPLE POLYPEPTIDE CHAINS)
- FUNCTIONS:
 - ENZYMATIC ACTIVITY
 - STRUCTURAL SUPPORT (E.G., COLLAGEN)
 - TRANSPORT (E.G., HEMOGLOBIN)

2.4. NUCLEIC ACIDS

- STRUCTURE: NUCLEIC ACIDS ARE POLYMERS MADE OF NUCLEOTIDE MONOMERS.
- TYPES:
 1. DNA (DEOXYRIBONUCLEIC ACID)
 2. RNA (RIBONUCLEIC ACID)
- FUNCTIONS:
 - GENETIC INFORMATION STORAGE AND TRANSMISSION (DNA)
 - PROTEIN SYNTHESIS (RNA)

3. CELL STRUCTURE AND FUNCTION

CELLS ARE THE BASIC UNIT OF LIFE, AND UNDERSTANDING THEIR STRUCTURE AND FUNCTION IS CRITICAL FOR ANY BIOLOGY STUDENT.

3.1. PROKARYOTIC VS. EUKARYOTIC CELLS

- PROKARYOTIC CELLS:
 - LACK MEMBRANE-BOUND ORGANELLES
 - SMALLER AND SIMPLER (E.G., BACTERIA)
 - GENETIC MATERIAL IS LOCATED IN THE NUCLEOID REGION

- EUKARYOTIC CELLS:
- CONTAIN MEMBRANE-BOUND ORGANELLES (E.G., NUCLEUS, MITOCHONDRIA)
- LARGER AND MORE COMPLEX (E.G., PLANT AND ANIMAL CELLS)

3.2. CELL ORGANELLES AND THEIR FUNCTIONS

- NUCLEUS: CONTAINS GENETIC MATERIAL (DNA) AND CONTROLS CELLULAR ACTIVITIES.
- RIBOSOMES: SITES OF PROTEIN SYNTHESIS.
- ENDOPLASMIC RETICULUM (ER):
- ROUGH ER: STUDED WITH RIBOSOMES; INVOLVED IN PROTEIN SYNTHESIS.
- SMOOTH ER: LACKS RIBOSOMES; INVOLVED IN LIPID SYNTHESIS AND DETOXIFICATION.
- GOLGI APPARATUS: MODIFIES, SORTS, AND PACKAGES PROTEINS AND LIPIDS FOR SECRETION OR USE WITHIN THE CELL.
- MITOCHONDRIA: THE POWERHOUSE OF THE CELL; SITE OF ATP (ENERGY) PRODUCTION.
- CHLOROPLASTS (IN PLANT CELLS): SITE OF PHOTOSYNTHESIS.
- CELL MEMBRANE: SEMI-PERMEABLE BARRIER THAT REGULATES WHAT ENTERS AND EXITS THE CELL.

3.3. CELL MEMBRANE STRUCTURE AND FUNCTION

- PHOSPHOLIPID BILAYER: COMPOSED OF TWO LAYERS OF PHOSPHOLIPIDS WITH HYDROPHILIC HEADS AND HYDROPHOBIC TAILS.
- EMBEDDED PROTEINS: INTEGRAL AND PERIPHERAL PROTEINS FACILITATE TRANSPORT AND COMMUNICATION.
- FLUID MOSAIC MODEL: DESCRIBES THE CELL MEMBRANE AS A DYNAMIC STRUCTURE WITH VARIOUS COMPONENTS MOVING FREELY.

4. ENERGY AND METABOLISM

UNDERSTANDING ENERGY TRANSFORMATIONS IN BIOLOGICAL SYSTEMS IS CRUCIAL FOR GRASPING HOW LIFE SUSTAINS ITSELF.

4.1. ATP (ADENOSINE TRIPHOSPHATE)

- STRUCTURE: COMPOSED OF ADENINE, RIBOSE, AND THREE PHOSPHATE GROUPS.
- FUNCTION: ATP IS THE PRIMARY ENERGY CURRENCY OF THE CELL, PROVIDING ENERGY FOR VARIOUS CELLULAR PROCESSES THROUGH HYDROLYSIS.

4.2. ENZYME FUNCTION

- ENZYMES: BIOLOGICAL CATALYSTS THAT SPEED UP CHEMICAL REACTIONS WITHOUT BEING CONSUMED.
- ACTIVE SITE: THE REGION ON THE ENZYME WHERE SUBSTRATES BIND.
- FACTORS AFFECTING ENZYME ACTIVITY:
- TEMPERATURE
- pH
- SUBSTRATE CONCENTRATION
- ENZYME CONCENTRATION

4.3. METABOLIC PATHWAYS

- ANABOLIC PATHWAYS: BUILD LARGER MOLECULES FROM SMALLER ONES (E.G., PROTEIN SYNTHESIS).

- CATABOLIC PATHWAYS: BREAK DOWN LARGER MOLECULES INTO SMALLER ONES (E.G., CELLULAR RESPIRATION).

5. STUDY STRATEGIES FOR AP BIOLOGY UNIT 1

TO EFFECTIVELY PREPARE FOR UNIT 1 OF AP BIOLOGY, CONSIDER THE FOLLOWING STUDY STRATEGIES:

- CREATE VISUAL AIDS: DIAGRAMS, FLOWCHARTS, AND CONCEPT MAPS CAN HELP VISUALIZE COMPLEX PROCESSES.
- PRACTICE MULTIPLE-CHOICE QUESTIONS: FAMILIARIZE YOURSELF WITH THE EXAM FORMAT AND QUESTION TYPES BY PRACTICING PAST AP QUESTIONS.
- USE FLASHCARDS: FOR MEMORIZING KEY TERMS AND DEFINITIONS, FLASHCARDS CAN BE AN EFFECTIVE TOOL.
- JOIN STUDY GROUPS: DISCUSSING TOPICS WITH PEERS CAN ENHANCE UNDERSTANDING AND RETENTION.
- UTILIZE ONLINE RESOURCES: WEBSITES LIKE KHAN ACADEMY, BOZEMAN SCIENCE, AND AP CLASSROOM PROVIDE VALUABLE VIDEOS AND PRACTICE QUESTIONS.

IN CONCLUSION, THE AP BIOLOGY UNIT 1 CHEAT SHEET SERVES AS A COMPREHENSIVE OVERVIEW OF THE FUNDAMENTAL CONCEPTS NECESSARY FOR SUCCESS IN THE COURSE. BY UNDERSTANDING THE CHEMISTRY OF LIFE, MACROMOLECULES, CELL STRUCTURE, AND ENERGY PROCESSES, STUDENTS WILL BE WELL-PREPARED TO TACKLE MORE ADVANCED TOPICS IN BIOLOGY. USE THIS CHEAT SHEET ALONGSIDE EFFECTIVE STUDY STRATEGIES TO MAXIMIZE YOUR LEARNING AND PERFORMANCE IN AP BIOLOGY.

FREQUENTLY ASKED QUESTIONS

WHAT TOPICS ARE COVERED IN THE AP BIOLOGY UNIT 1 CHEAT SHEET?

THE AP BIOLOGY UNIT 1 CHEAT SHEET TYPICALLY COVERS KEY CONCEPTS SUCH AS THE PROPERTIES OF WATER, MACROMOLECULES, CELL STRUCTURE AND FUNCTION, AND THE BASICS OF CELLULAR PROCESSES.

HOW CAN I EFFECTIVELY USE A CHEAT SHEET FOR AP BIOLOGY UNIT 1?

YOU CAN USE A CHEAT SHEET BY SUMMARIZING KEY TERMS, CONCEPTS, AND DIAGRAMS, WHICH HELPS IN QUICK REVISION AND REINFORCES YOUR UNDERSTANDING OF THE MATERIAL.

WHAT IS THE IMPORTANCE OF WATER IN BIOLOGICAL SYSTEMS ACCORDING TO UNIT 1?

WATER IS CRUCIAL IN BIOLOGICAL SYSTEMS AS IT ACTS AS A SOLVENT, HELPS REGULATE TEMPERATURE, AND PARTICIPATES IN BIOCHEMICAL REACTIONS, MAKING IT ESSENTIAL FOR LIFE.

WHAT MACROMOLECULES ARE HIGHLIGHTED IN AP BIOLOGY UNIT 1?

AP BIOLOGY UNIT 1 HIGHLIGHTS FOUR MAIN TYPES OF MACROMOLECULES: CARBOHYDRATES, LIPIDS, PROTEINS, AND NUCLEIC ACIDS, INCLUDING THEIR STRUCTURES AND FUNCTIONS.

ARE THERE ANY DIAGRAMS THAT ARE ESSENTIAL FOR UNIT 1 IN THE CHEAT SHEET?

YES, DIAGRAMS SUCH AS THE CELL STRUCTURE, THE FLUID MOSAIC MODEL OF THE CELL MEMBRANE, AND METABOLIC PATHWAYS ARE ESSENTIAL FOR VISUALIZING CONCEPTS IN UNIT 1.

WHAT STRATEGIES CAN I USE TO MEMORIZE KEY CONCEPTS FROM UNIT 1?

STRATEGIES INCLUDE USING FLASHCARDS, CREATING MNEMONIC DEVICES, PRACTICING WITH QUIZZES, AND DISCUSSING TOPICS WITH PEERS TO REINFORCE MEMORY.

How does AP Biology Unit 1 relate to the overall AP Biology curriculum?

Unit 1 lays the foundation for understanding the biochemical and cellular basis of life, which is essential for grasping more complex topics in subsequent units of the AP Biology curriculum.

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